

Ink-Jet Printing of Concentrated Polymer Solutions

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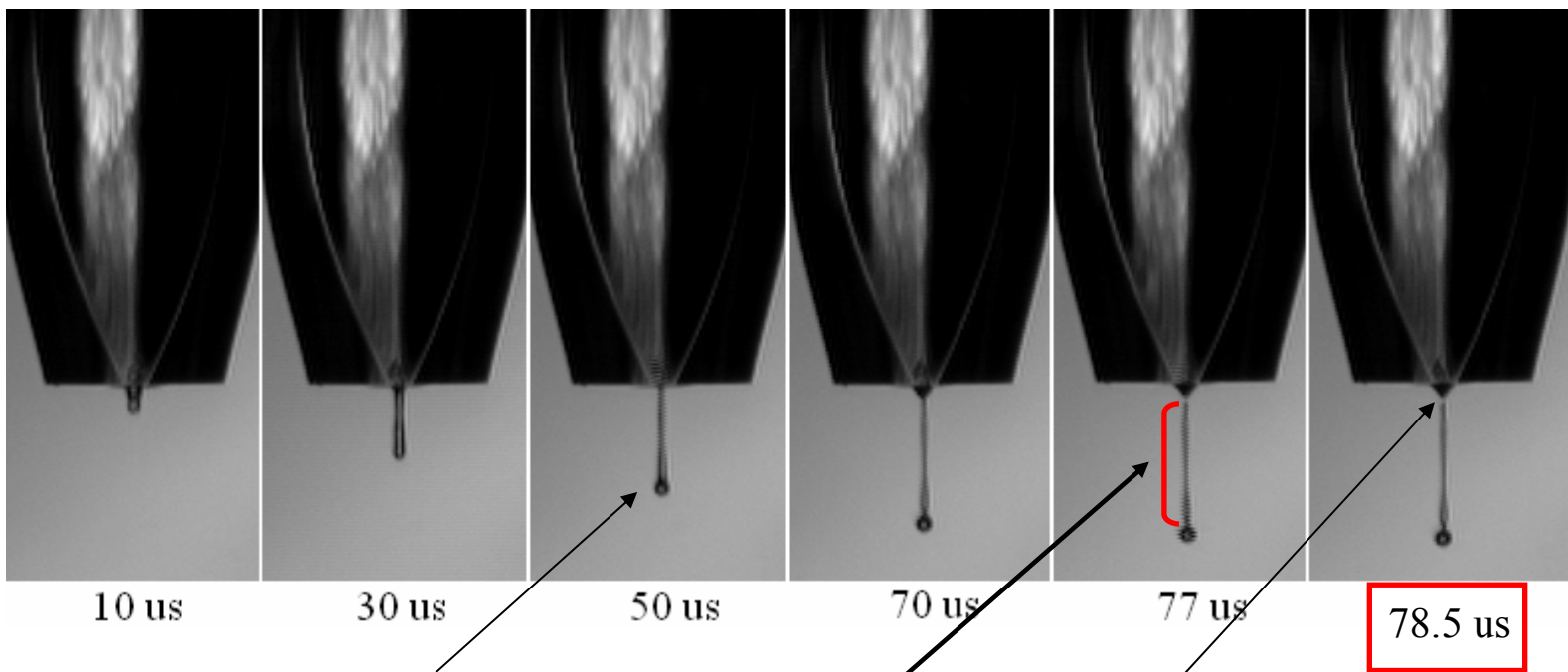
Talk

- **Motivation**
- **Experimental**
- **Effect of molecular weight at overlap concentration**
- **Effect of polymer concentration through the dilute to semi-dilute and concentrated regime**
- **Some conclusions**

Motivation

- The use of **HMW polymer additives** is well established in inkjet for use in **drag reduction** and **ligament control** at low concentration.
- Demands of graphics, industrial and emerging applications require the **deposition of functional polymers** at increased concentrations.
- Here we explore the inkjet behaviour of **cellulose esters** as a model system to look at the **effects of molecular weight and concentration** for comparatively low molar mass functional polymers.

Strobed photo-sequence of a 'typical' polymer fluid



Primary droplet

Ligament

Pinch off

Rupture time (t_{rup})

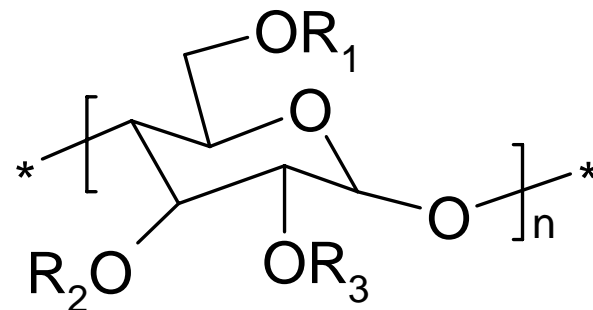
Materials ...

Cellulose Ester Polymers

CE	Composition (wt-%)		Hydroxy	M _n (kDa)	PDI
	Acetyl	Butyryl			
1	2	53	1	10.5	2.90
2	2	52	1	30.7	2.49
3	3	50	1	46.5	3.39

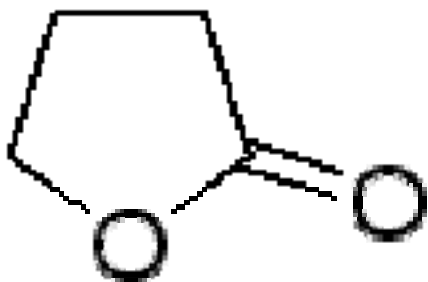
Advantages

1. Excellent solubility;
2. Comparatively low viscosity;
3. Reduced drying time;
4. Good flow;
5. Good pigment dispersion medium.



Solution Properties: γ -Butyrolactone

CE	$[\eta]$ (dl/g)	c^*	λz (μs)	c (g/dl)	c/c^*	η (cP)	γ (mN/m)
1	0.247	4.04	0.53	4.0	1.0	4.2	33.0
2	0.631	1.58	3.43	0.5	0.3	2.2	30.1
				1.0	0.6	2.7	30.6
				1.5	1.0	3.6	31.2
				2.0	1.3	4.7	31.5
				4.0	2.5	11.5	34.0
				6.0	3.8	20.3	35.6
3	1.153	0.86	12.9	0.86	1	4.7	35.0



γ -Butyrolactone

Concentration Regimes

- **Dilute:** Polymer chains are at a concentration at which the polymer chains are **isolated**. Solution concentration rises very slowly with increasing polymer concentration up to the overlap concentration.
- **Semi-dilute:** Polymer chains are at a concentration at which they **just overlap**. This point is called the overlap concentration c^* .

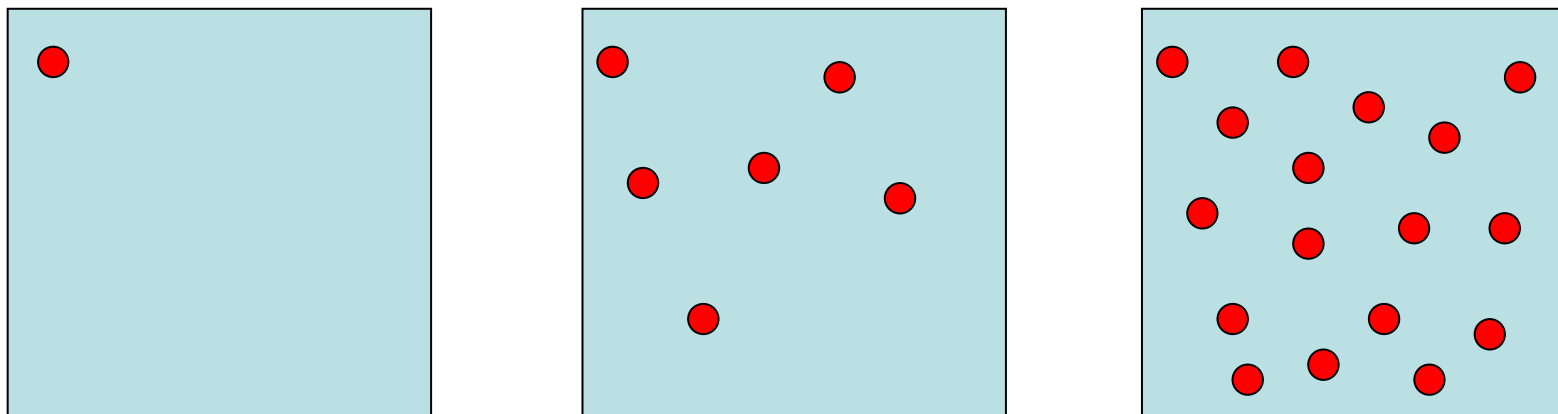
$$C^* = 1/[\eta]$$

- **Concentrated:** Polymer chains **overlap** and viscosity of the solution rises steeply with polymer concentration.

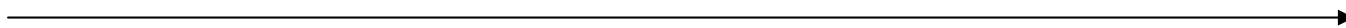
C/C^* : reduced concentration

Details ...

Dilute polymer solution

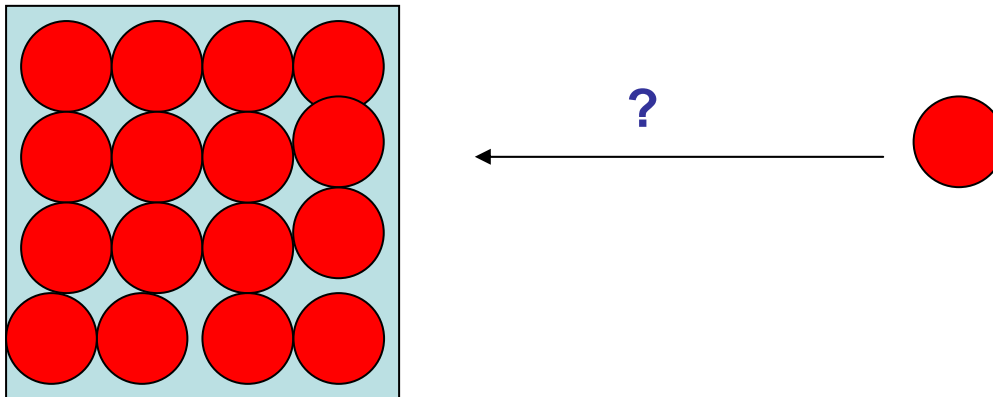


Increasing polymer Concentration



Elasticity of polymer solution increased

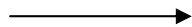
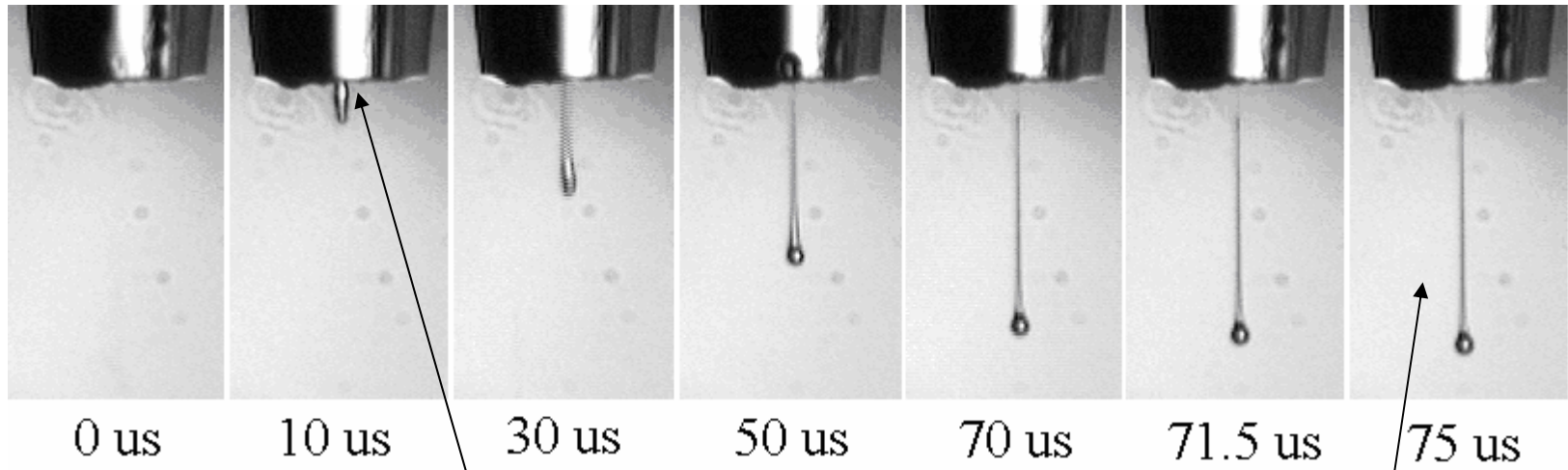
Semi-dilute polymer solution



The overlap concentration $c^* = 1/[\eta]$

Polymer chains' behaviour
during Ink-jetting ...

Complex Fluids Under High Shear



Relaxation time (τ_1)



- Effect of polymer structure
- Dilute/semi dilute

Relaxation time (τ_1) ...

For dilute polymer solutions the longest relaxation time is given by the Zimm, non free draining **relaxation time** (τ_1):

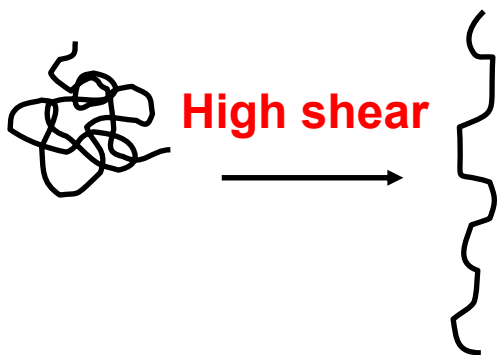
$$\tau_1 = \eta_s [\eta] M / RT$$

where η_s is the viscosity of the solvent, $[\eta]$ the intrinsic viscosity of the polymer solution and M the number average molecular weight.

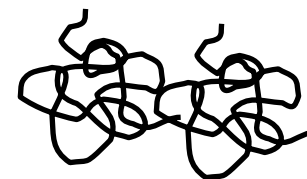
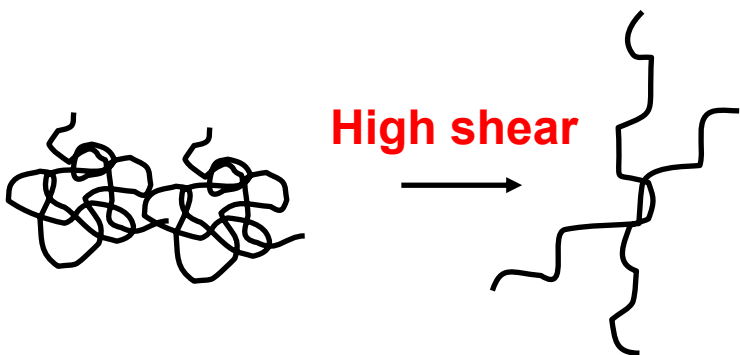
When the **strain is switched off** the **extended** polymer **returns** to the more thermodynamically stable **coil state**. It does this slowly because the extended stable chain experiences the full viscous drag of the solvent.

Chain Entanglement During inkjet process

$c/c^* \leq 1$



$c/c^* > 1$



overlap

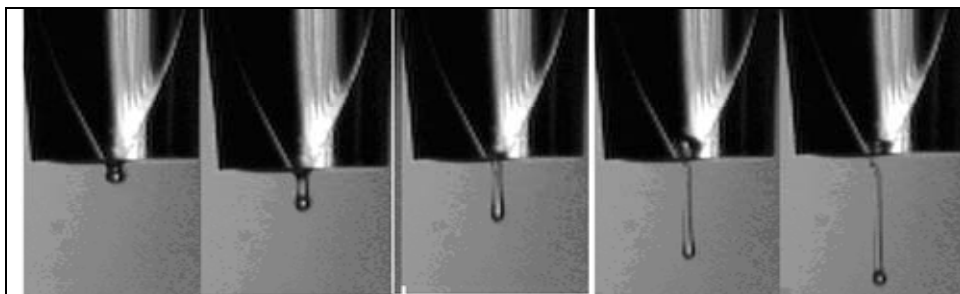
entangled

Need more time to come
back to coil state

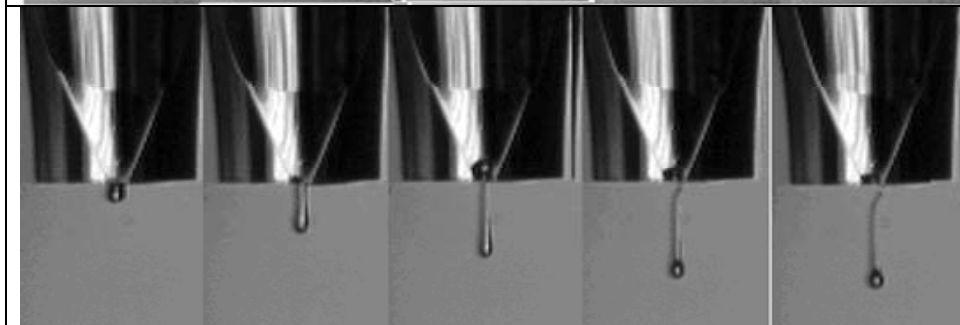
Effect of M_w ...

Effect of Molecular Weight at $c/c^* = 1$

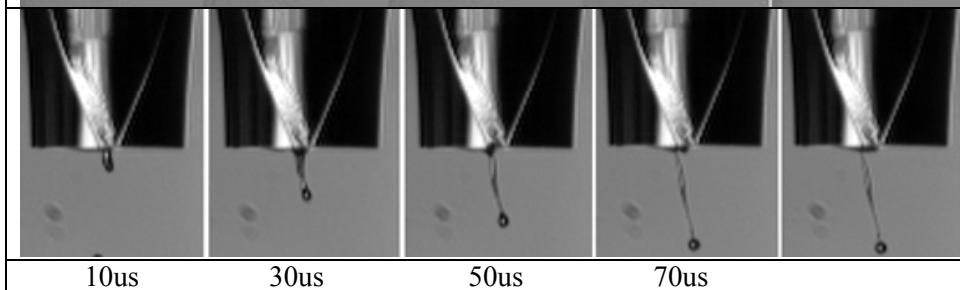
CE1 (4)



CE2 (1.5)



CE3 (0.9)



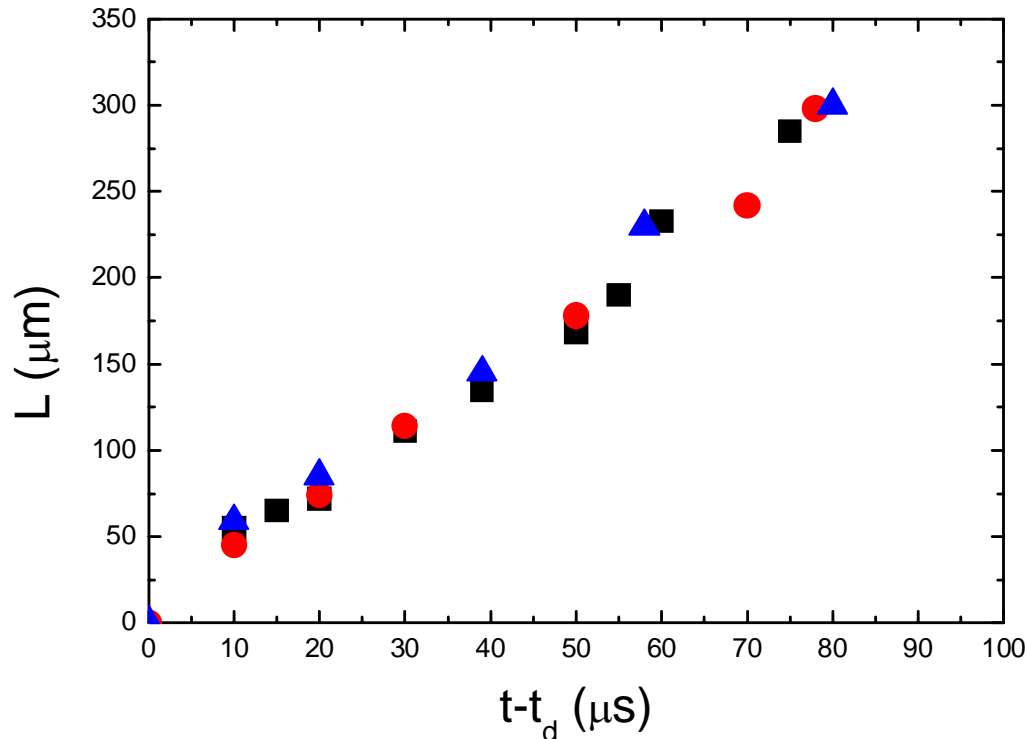
10us

30us

50us

70us

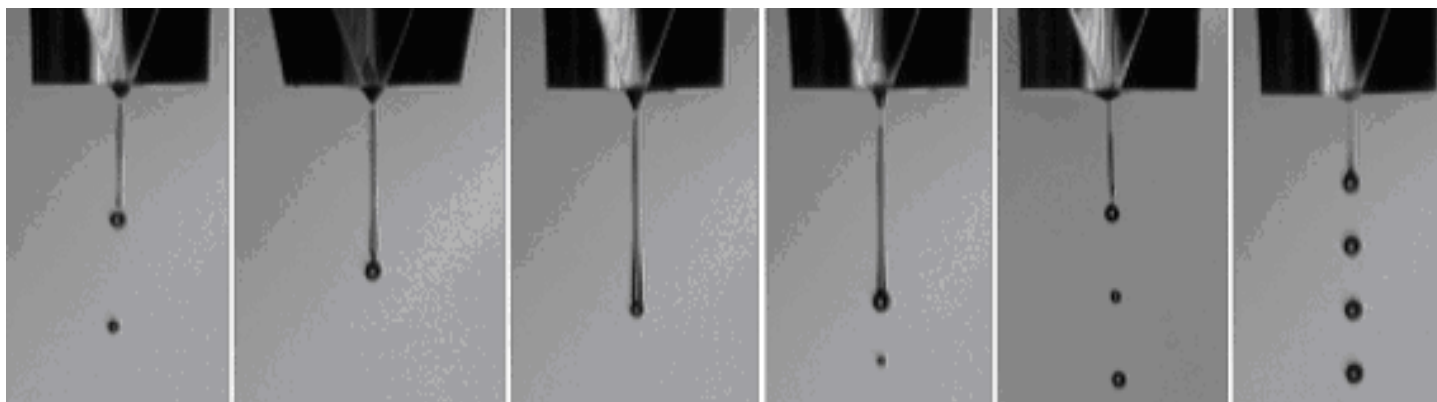
Effect of Molecular Weight at $c/c^* = 1$



Ligament length at rupture, L_{rup} , as a function of rupture time, $t_{rup} = t-t_d$.

- Formulated at c/c^* show essentially similar behaviour;
- Same ejection velocities for a given waveform.

CE2 Effect of Concentration



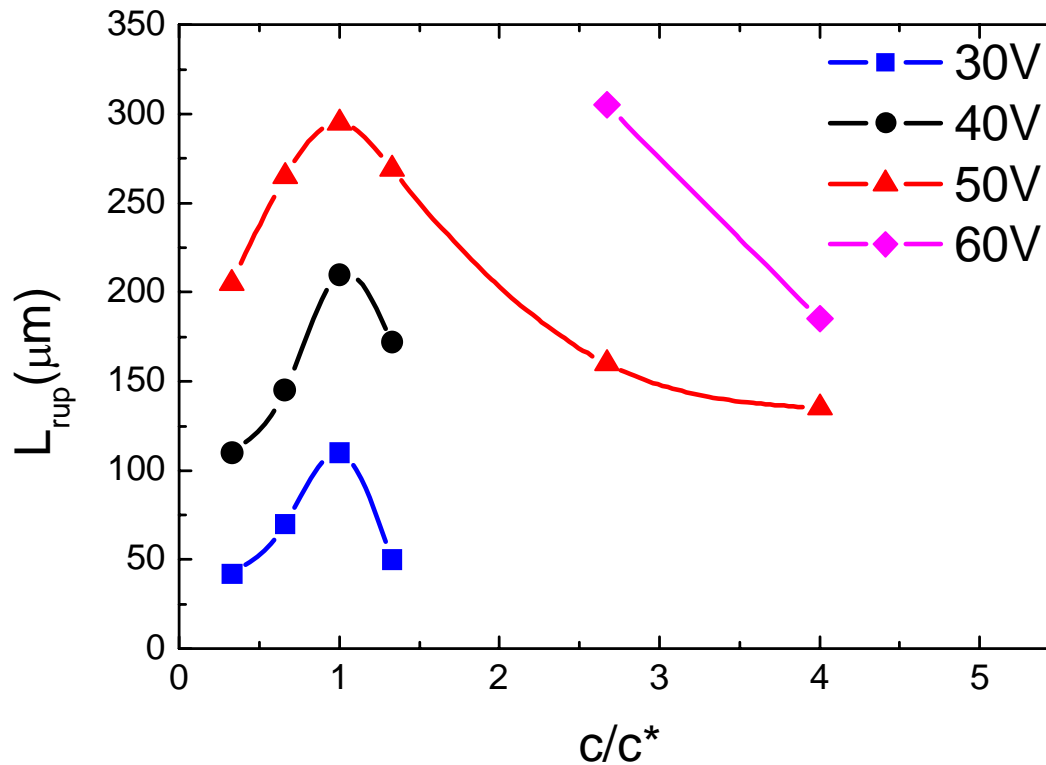
c/c^* 0.32 0.63 0.96 1.28 2.52 3.84

Drop profile at various concentration (applied voltage = 50v).

➤ c/c^* : reduced concentration;

➤ It is observed that the **ligament length at rupture increases** up to a reduced concentration of 1 and **decreases** at higher concentration.

Ligament Length

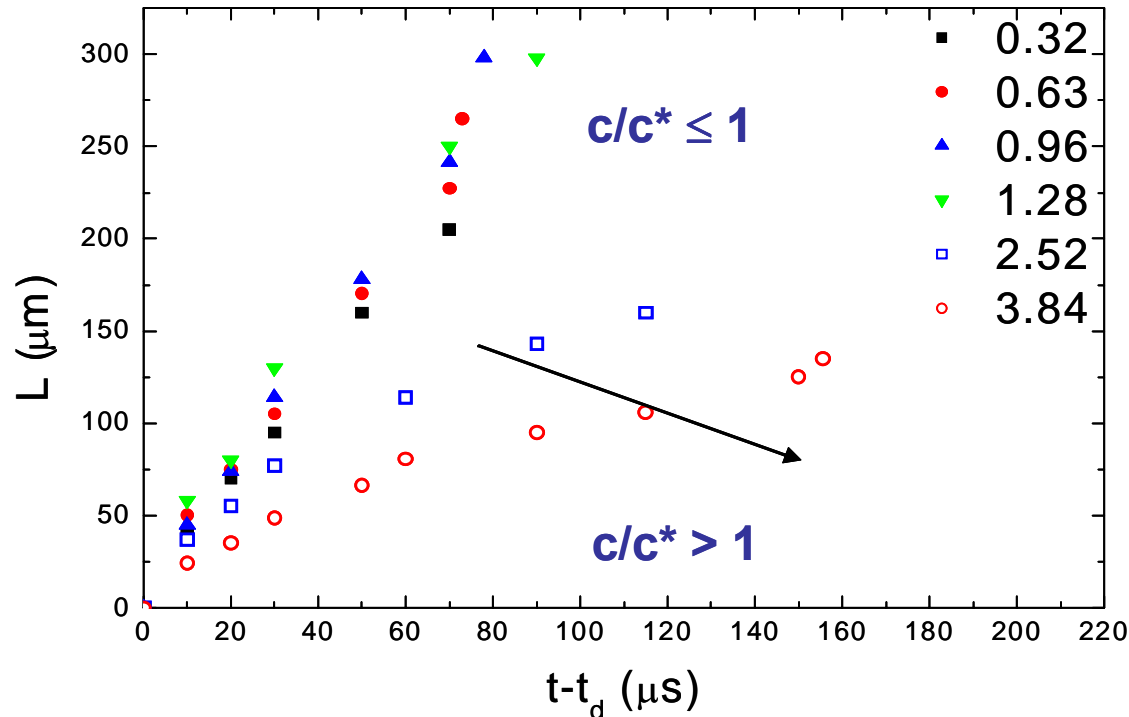


Ligament length at rupture, L_{rup} , as a function of reduced concentration c/c^* .

- It shows the influence of reduced concentration and applied voltage (30 - 60 V) on the ligament length at rupture (L_{rup});
- Maximum L_{rup} is at around the overlap concentration;
- L_{rup} decreased rapidly at higher concentrations ($c/c^* > 1$);
- Due to chain entanglement.

L_{rup} Vs. t_{rup} ...

Ligament length as a function of $t - t_d$



For an applied voltage of 50 V, the **travelled distance increases** linearly with time and the velocity becomes constant.

- $t - t_d < 20 \mu\text{s}$: the trace is **independent** of polymer **C** and **Mw**;
Interplay of inertia Vs. ST.
- $c/c^* \leq 1$: $L-t$ data are found to collapse to a **single curve**;
- $c/c^* > 1$: the slope of the curve becomes much **shallower** and is highly dependant upon concentration.

Observations ...

Observations

$$c/c^* \leq 1$$

$$c/c^* > 1$$

Increasing concentration:

- Rupture time and droplet velocity dominated by **applied voltage**.

- ligament/droplet **velocity decreases** significantly;
- **Ligaments are shorter and take longer to break**;
- **If concentration too high no break off**.

Conclusions

- We can now define a series of design **rules for the inkjet deposition** of polymer containing fluids.
- Want τ_1 to be as small as possible so that viscosity increase through hydrodynamic drag is minimized.
- Can be achieved by:
 - Good solvent for whole polymer chain.
 - No aggregation in solution.
 - Formulate where possible below C^* .
 - Lower molecular weight the better.

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